

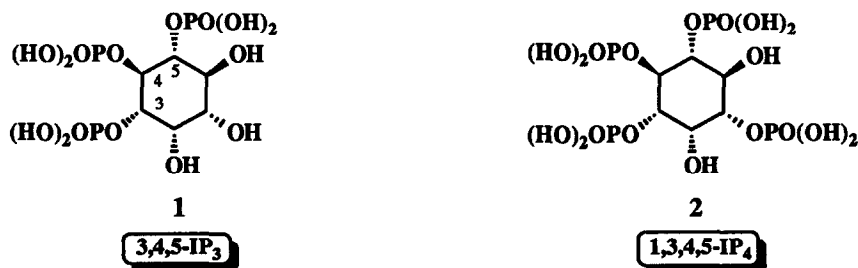
TOTAL SYNTHESIS OF D-MYO-INOSITOL 3,4,5-TRISPHOSPHATE AND 1,3,4,5-TETRAKISPHOSPHATE

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Abstract. A differentially protected chiral cyclitol was prepared from (-)-quinic acid and exploited for the total synthesis of the title *myo*-inositol polyphosphates.

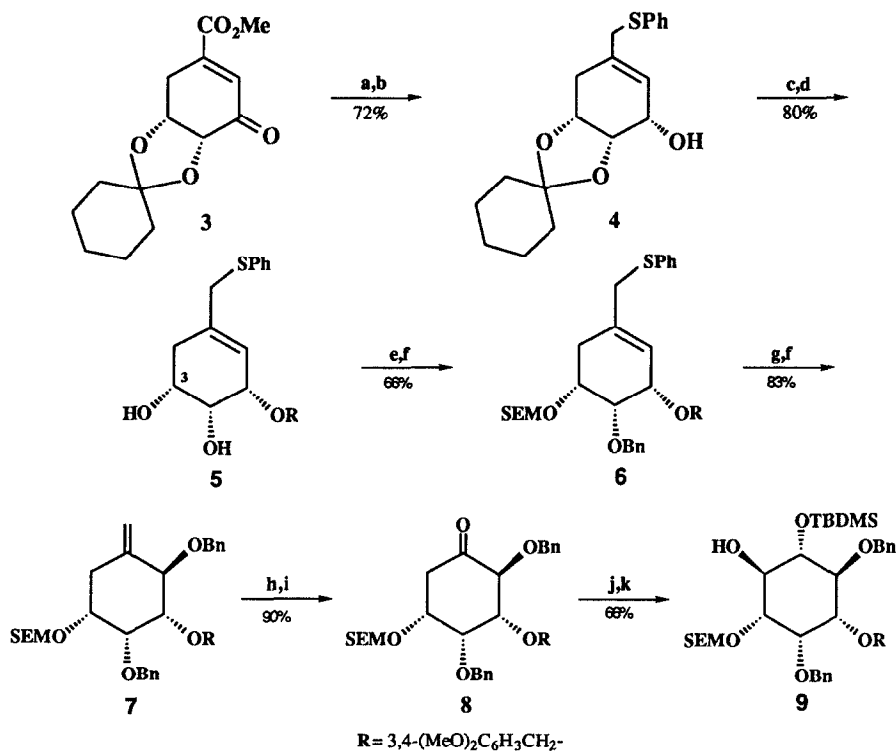
The extensive investigations during recent years of the phosphatidylinositol (PI) cycle have revealed nearly two dozen inositol phosphates and related glycerylphospholipids.¹ The more recent discovery of a metabolically distinct class of 3-phosphorylated PIs associated with cellular proliferation² portends even greater structural diversity. While the complex interrelationship amongst many of the inositol metabolites has been established¹, the physiological role for all but a few remains obscure due, in part, to their limited availability from natural sources.³ To help encompass the burgeoning number of new metabolites and to satisfy the urgent requirement for isomerically homogeneous standards, we report herein a significant extension of our earlier approach⁴ to functionalized, chiral cyclitols from (-)-quinic acid and illustrate its versatility during total synthesis of D-*myo*-inositol 3,4,5-trisphosphate (1) and 1,3,4,5-tetrakisphosphate (2).



Enone 3, obtained in 68% yield from (-)-quinic acid⁴, was converted to phenylsulfide 4⁵ (Scheme I) by concurrent diisobutylaluminum hydride (DIBAL-H) reduction of the ester and keto groups followed by selective replacement of the primary alcohol using phenyl disulfide/tributylphosphine.⁶ Protection of the remaining secondary alcohol as its 3,4-dimethoxybenzyl ether and mild acidic hydrolysis of the cyclohexylidene furnished diol 5⁷, mp 85-86°C. Preferential etherification of the equatorial C(3)-alcohol (inositol numbering) via the corresponding cyclic stannylene acetal⁸, prepared *in situ* from 5 and dibutyltin oxide, using equimolar amounts of 2-(trimethylsilyl)ethoxymethyl (SEM) chloride and CsF^{9,10}, then conventional benzylation of the axial C(2)-alcohol led to 6. Low temperature peracid oxidation of 6 gave an ~1:1 mixture of sulfoxides that was transformed into olefin 7, free of any α -isomer, by carefully controlled [2,3]-sigmatropic rearrangement⁴ at 40°C and benzylation of the newly generated allylic alcohol. Since the 3,4-dimethoxybenzyl ether did not survive ozonolytic cleavage of the exocyclic olefin, ketone 8 was secured by an efficient two-step sequence involving OsO₄ glycolization and Pb(OAc)₄ oxidation. Regioselective conversion to silyl enol ether using excess *tert*-butyldimethylsilyl (TBDMS) triflate¹¹ and hydroboration

exclusively from the more accessible β -face was rewarded, after peracid work-up, by the fully hydroxylated and differentially protected cyclitol **9**.

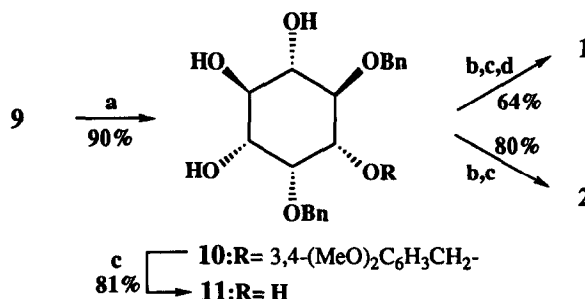
Scheme I



^aDIBAL-H, PhCH₃, -78°C, 1h. ^bPhSSPh, Bu₃P, 0°C, 0.5h. ^cNaH, RBr, THF, 24°C, 2h. ^d1N HCl (1.5 equiv), MeOH, 5°C, 60h. ^e*n*-Bu₂SnO, PhCH₃, 120°C, 4h; SEM-Cl, CsF, CH₂Cl₂, -45°C, 12h. ^fNaH, BnBr, THF, 8h. ^g*m*-CPBA, CH₂Cl₂, -78°C, 0.5h; (EtO)₃P (5 equiv), MeOH, 40°C, 11d. ^hOsO₄, NMO, H₂O/acetone/*t*-BuOH (3:3:1), 0°C, 1h, then 24°C, 4h. ⁱPb(OAc)₄, PhH, 5°C, 20 min. ^jEt₃N (15 equiv), TBDMSOTf (6 equiv), CH₂Cl₂, 0→18°C, 4h. ^kBH₃·THF, CH₂Cl₂, 5°C, 5d; H₂O₂/NaOH; *m*-CPBA.

The utility of **9** for the synthesis of inositol polyphosphates was demonstrated (Scheme II) by hydrolysis of the SEM and silyl ethers to give triol **10**, mp 142.5 - 143.5°C. Phosphorylation (80%) according to the method¹² of Tegge and Ballou, DDQ promoted¹³ cleavage of the 3,4-dimethoxybenzyl ether (89%), and catalytic hydrogenolysis (10% Pd/C, 50 psig H₂, 80% aq. EtOH) of the benzyl protecting groups gave rise to 3,4,5-IP₃ (**1**) (90%), isolated as the sodium salt. This is the first chemical synthesis of **1** which has been isolated from rat mammary tumor cells after stimulation¹⁴ and avian erythrocytes.¹⁵ Alternatively, treatment of **10** with DDQ afforded tetraol **11** as colorless crystals (81%), mp 144.5-145.5°C, [α]_D²⁰⁻²⁷(c 0.32, EtOH); lit.^{16a} mp 145.2-146.1°C, [α]_D^{25-29.2}(c 1.0, EtOH). Phosphorylation and hydrogenolysis as above gave the calcium mobilizing second messenger¹⁷ 1,3,4,5-IP₄¹⁶⁽²⁾ (80%), also isolated as the sodium salt.

Scheme II



^a1N HCl (2 equiv), MeOH, 45°C, 1.5h. ^b(i-Pr)₂NP(OBn)₂, 1-H-tetrazole, CH₂Cl₂, 15°C, 1.5 h; mCPBA, -78→0°C, 1.5 h. ^cDDQ (1.5 equiv), CH₂Cl₂/H₂O (20:1), 0°C, 1.2h. ^dH₂, 10% Pd/C, 50 psig, 85% aq EtOH; NaOH.

The exploitation of **9** for the preparation of phosphatidylinositol polyphosphates and the results of pharmacologic evaluations from **1** will be reported elsewhere.

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References and Notes

- Reviews: Downs, C.P.; Macphee, C.H. *Eur. J. Biochem.* **1990**, *193*, 1-18. Joseph, S.K.; Williamson, J.R. *Arch. Biochem. Biophys.* **1989**, *273*, 1-15. Fisher, S.K.; Heacock, A.M.; Agranoff, B.W. *J. Neurochem.* **1992**, *58*, 18-38.
- Pignataro, O.P.; Ascoli, M. *J. Biol. Chem.* **1990**, *265*, 1718-1723. Auger, K.R.; Carpenter, C.L.; Cantley, L.C.; Varticovski, L. *ibid.* **1989**, *264*, 20181-20184.
- Reviews of chemical synthesis: Potter, B.V.L. *Nat. Prod. Reports* **1990**, *7*, 1-24. Billington, D.C. *Chem. Soc. Rev.* **1989**, *18*, 83-122.
- Falck, J.R.; Yadagiri, P. *J. Org. Chem.* **1989**, *54*, 5851-5852. Falck, J.R.; Abdali, A.; Wittenberger, S.J. *J. Chem. Soc., Chem. Commun.* **1990**, 953-955.
- Satisfactory spectral data were obtained for all new compounds using chromatographically homogeneous samples.
- Nakagawa, I.; Hata, T. *Tetrahedron Lett.* **1975**, 1409-1412.
- ¹H NMR (CDCl₃, 250 MHz) of **5**: δ 2.47 (d, J~5.7 Hz, 2H), 2.67 (d, J~5.7 Hz, OH), 2.71 (d, J~9.9 Hz, OH), 3.45 (d, J~13.8 Hz, 1H), 3.50 (d, J~13.8 Hz, 1H), 3.81 - 3.95 (complex m with 2 x OCH₃ at 3.87 and 3.88, 9H), 4.42 (d, J~11.4 Hz, 1H), 4.50 (d, J~11.4 Hz, 1H), 5.50 (br s, 1H), 6.81 (s, 3H), 7.16-7.36 (m, 5H); [α]_D²⁰ + 30.2° (c 1.1, CH₃CN). **7**: δ 0.02 (s, 9H), 0.84-0.96 (m, 2H), 2.42 (dd, J~4.6, 12.8 Hz, 1H), 2.63 (apparent t, J~11.6 Hz, 1H), 3.31 (dd, J~2.4, 9.4 Hz, 1H), 3.50-3.68 (complex m with OCH₃ singlet at 3.65, 6H), 3.83 (s, 3H), 4.03 (br s, 1H), 4.32 (d, J~9.4 Hz, 1H), 4.54-4.74 (m, 6H), 4.82 (d, J~11.6 Hz, 1H), 4.89 (d, J~11.6 Hz, 1H), 4.95 (br s, 1H), 5.19 (br s, 1H), 6.80-6.86 (m, 3H), 7.21-7.39 (m, 10H);

- $[\alpha]^{20}_D$ -51° (c 0.73, CH₃CN). **9**: δ 0.00 (s, 9H), 0.03 (s, 3H), 0.10 (s, 3H), 0.85-1.00 (m with *t*-Bu singlet at 0.88, 11H), 2.66 (d, *J*~2.4 Hz, OH), 3.36 (ddd, *J*~9.8, 9.8, 2.4 Hz, 1H), 3.42-3.91 (complex m with 2 x OCH₃ singlets at 3.57 and 3.86, 12H), 4.06 (t, *J*~2.3 Hz, 1H), 4.53 (d, *J*~11 Hz, 1H), 4.58 (d, *J*~11 Hz, 1H), 4.69-4.99 (m, 6H), 6.74-6.78 (m, 3H), 7.22-7.42 (m, 10H); $[\alpha]^{20}_D$ -38.3° (c 1.2, CH₃CN). **10**: δ 2.35 (d, *J*~8.2 Hz, OH), 2.61 (br s, OH), 2.74 (br s, OH), 3.38 (apparent t, *J*~9.6, 7.8 Hz, 2H), 3.48 (dd, *J*~2.3, 9.7 Hz, 1H), 3.70-3.88 (complex m with 2 x OCH₃ at 3.70 and 3.88, 8H), 4.04 (br s, 1H), 4.65 (s, 2H), 4.69 (d, *J*~11.6 Hz, 1H), 4.76 (d, *J*~11.3 Hz, 1H), 5.04 (apparent t, *J*~11.2, 10.5 Hz, 2H), 6.80-6.89 (m, 3H), 7.28-7.36 (m, 10H); $[\alpha]^{20}_D$ -39.0° (c 0.6, CH₃CN). **1** (hexasodium salt; D₂O, 500 MHz): δ 3.63 (dd, *J*~2.2, 9.0 Hz, 1H), 3.87-3.92 (m, 2H), 4.00 (brt, *J*~7.5 Hz, 1H), 4.36 (q, *J*~9.0 Hz, 1H), 4.42 (t, *J*~2.8 Hz, 1H).
8. David, S.; Hanessian, S. *Tetrahedron* **1985**, *41*, 643-663.
 9. Nagashima, N.; Ohno, M. *Chem. Lett.* **1987**, 141-144.
 10. CsF catalysis gave a 24:1 ratio of C-3 (equatorial)/C-2 (axial) SEM ethers whereas *n*-Bu₄NF showed lower selectivity (4:1).
 11. Corey, E.J.; Cho, H.; Rucker, C.; Hua, D.H. *Tetrahedron Lett.* **1981**, *22*, 3455-3458.
 12. Tegge, W.; Ballou, C.E. *Proc. Natl. Acad. Sci. USA* **1989**, *86*, 94-98.
 13. Oikawa, Y.; Tanaka, T.; Horita, K.; Yoshioka, T.; Yonemitsu, O. *Tetrahedron Lett.* **1984**, *25*, 5393-5396.
 14. Wong, N.S.; Barker, C.J.; Morris, A.J.; Craxton, A.; Kirk, C.J.; Michells, R.H. *Biochem.* **1992**, *286*, 459-468.
 15. Stephens, L.R.; Hawkins, P.T.; Downes, C.P. *Biochem. J.* **1989**, *262*, 727-737.
 16. Asymmetric syntheses of **2**: (a) Baudin, G.; Glanzer, B.I.; Swaminathan, K.S.; Vasella, A. *Helv. Chim. Acta* **1988**, *71*, 1367-1378. (b) Watanabe, Y.; Oka, A.; Shimizu, Y.; Ozaki, S. *Tetrahedron Lett.* **1990**, *31*, 2613-2616. (c) Watanabe, Y.; Fujimoto, T.; Shinohara, T.; Ozaki, S. *J. Chem. Soc., Chem. Commun.* **1991**, 428-429. Gou, D.-M.; Chen, C.-S. *Tetrahedron Lett.* **1992**, *33*, 721-724.
 17. Gawler, D.J.; Potter, B.V.L.; Nahorski, S.R. *Biochem J.* **1990**, *272*, 519-524.